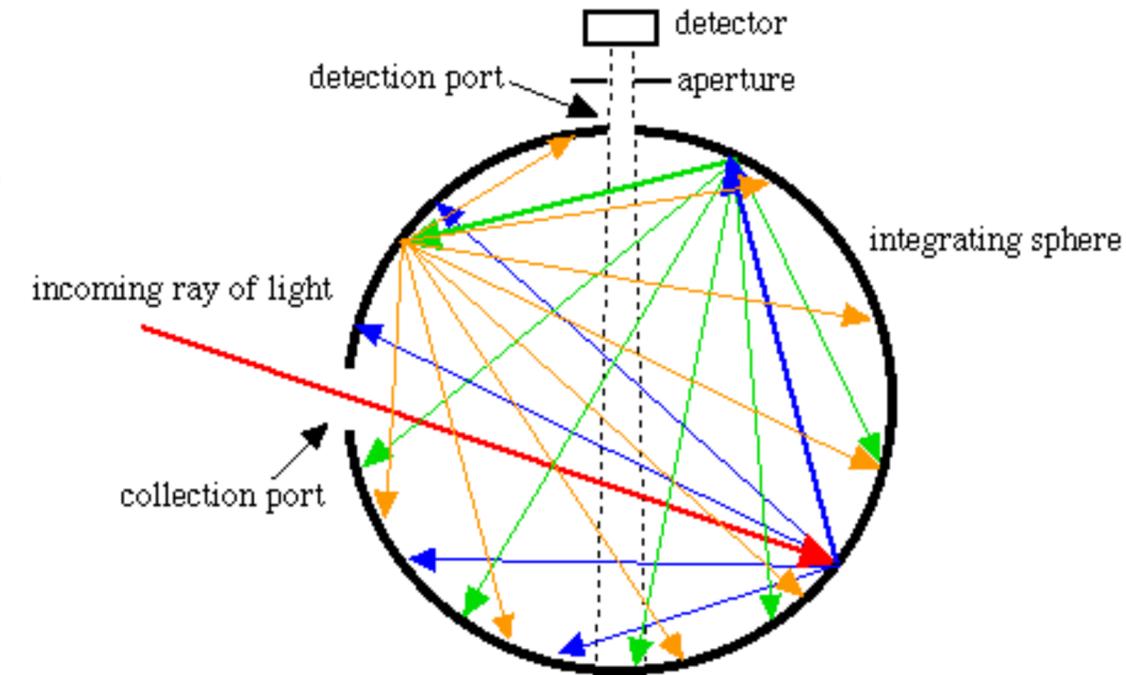
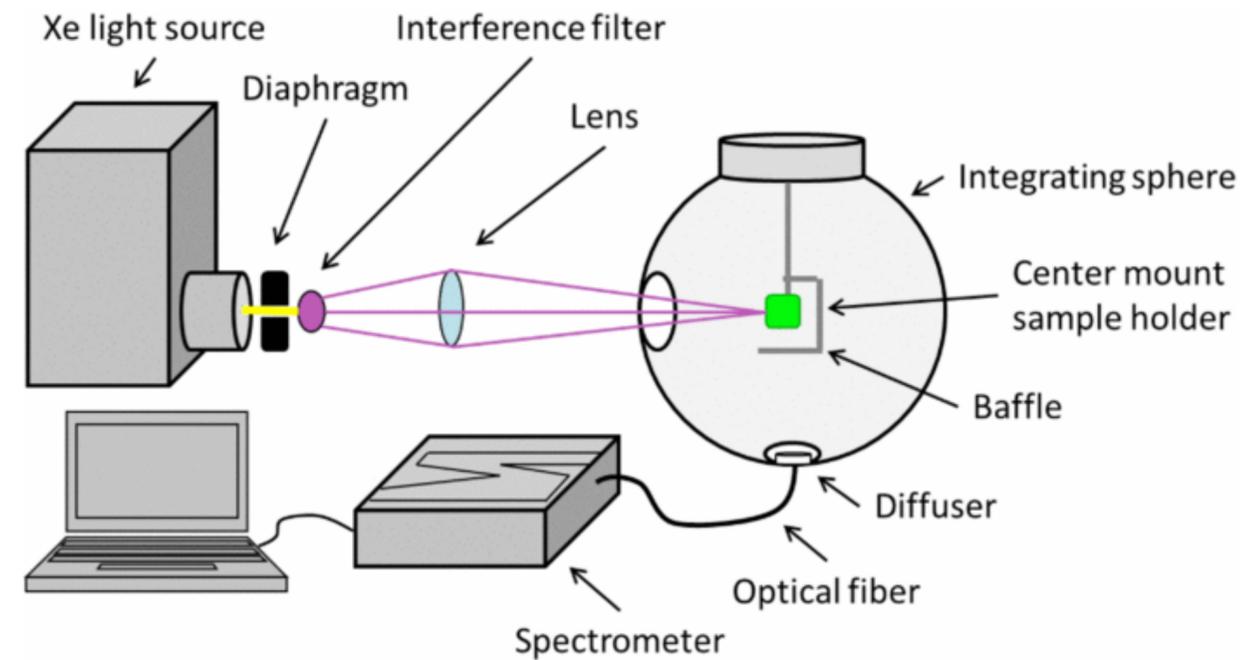
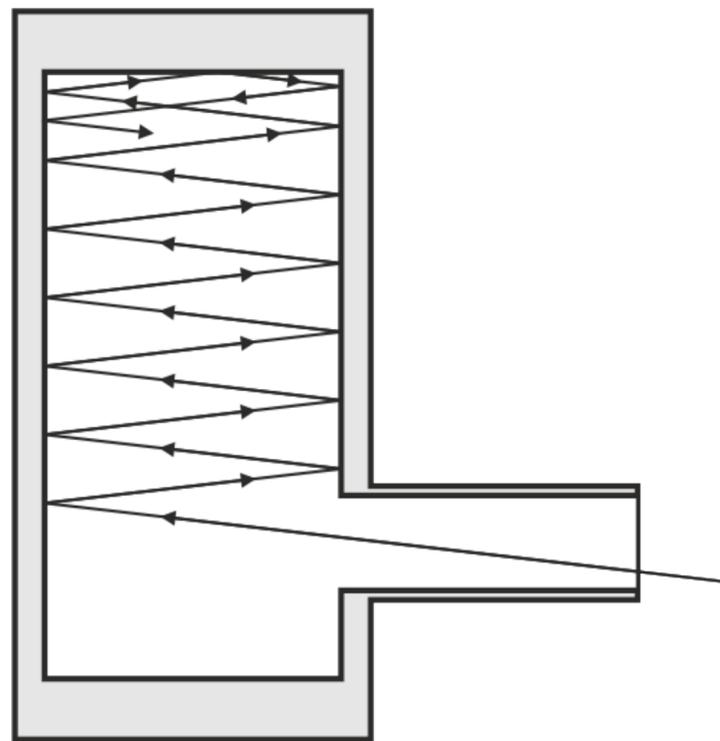


Experimental Physics 3 - Em-Waves, Optics, Quantum mechanics

Lecture 21

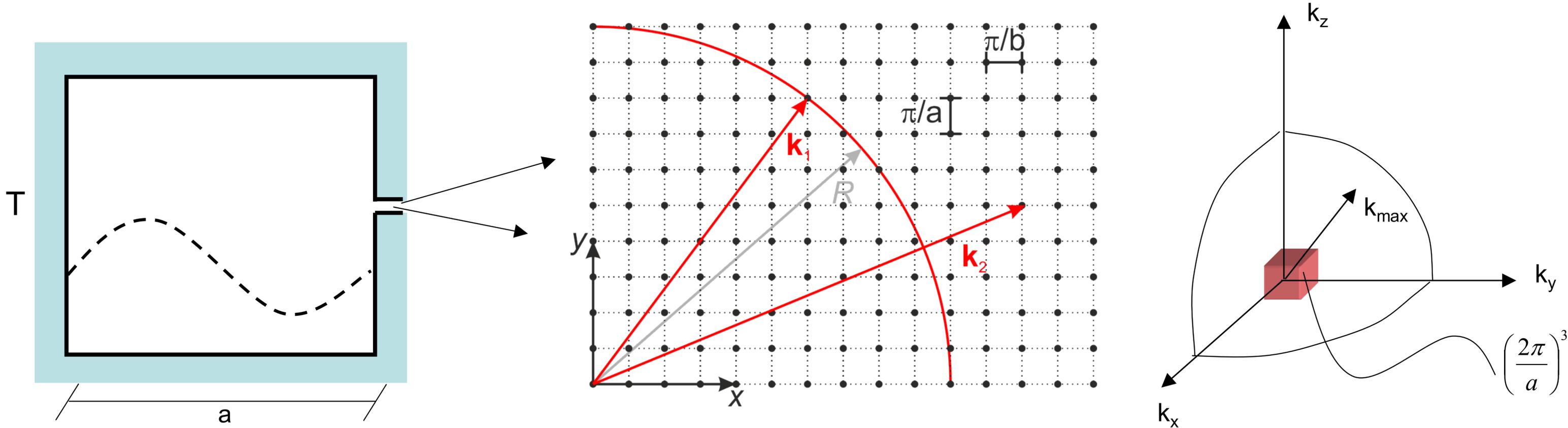
The Birth of Quantum Mechanics

Cavity radiator and integrating or Ulbricht sphere

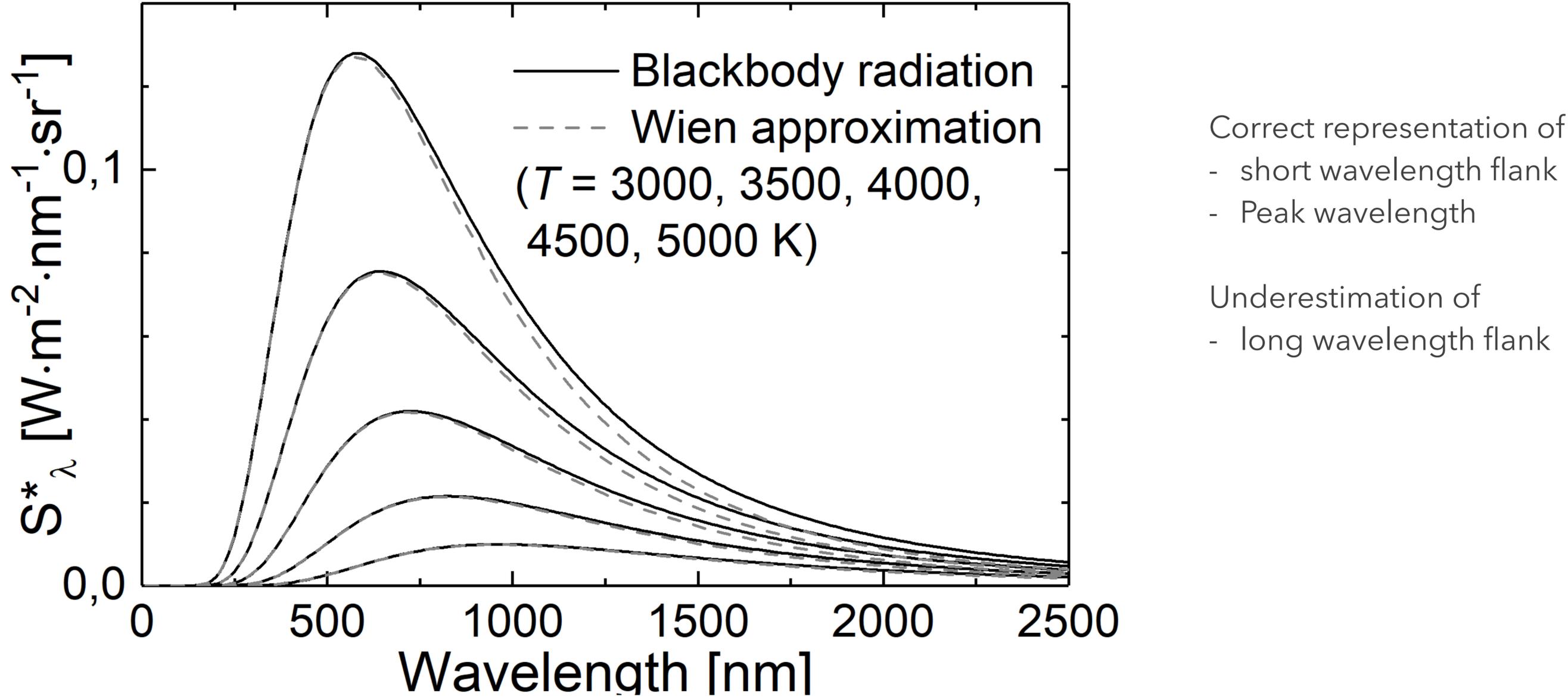


- Integrating sphere or Ulbricht sphere
- Total radiant flux collection of
 - emitted and scattered by sample
 - entered through inlet port
 - Direction-independent collection

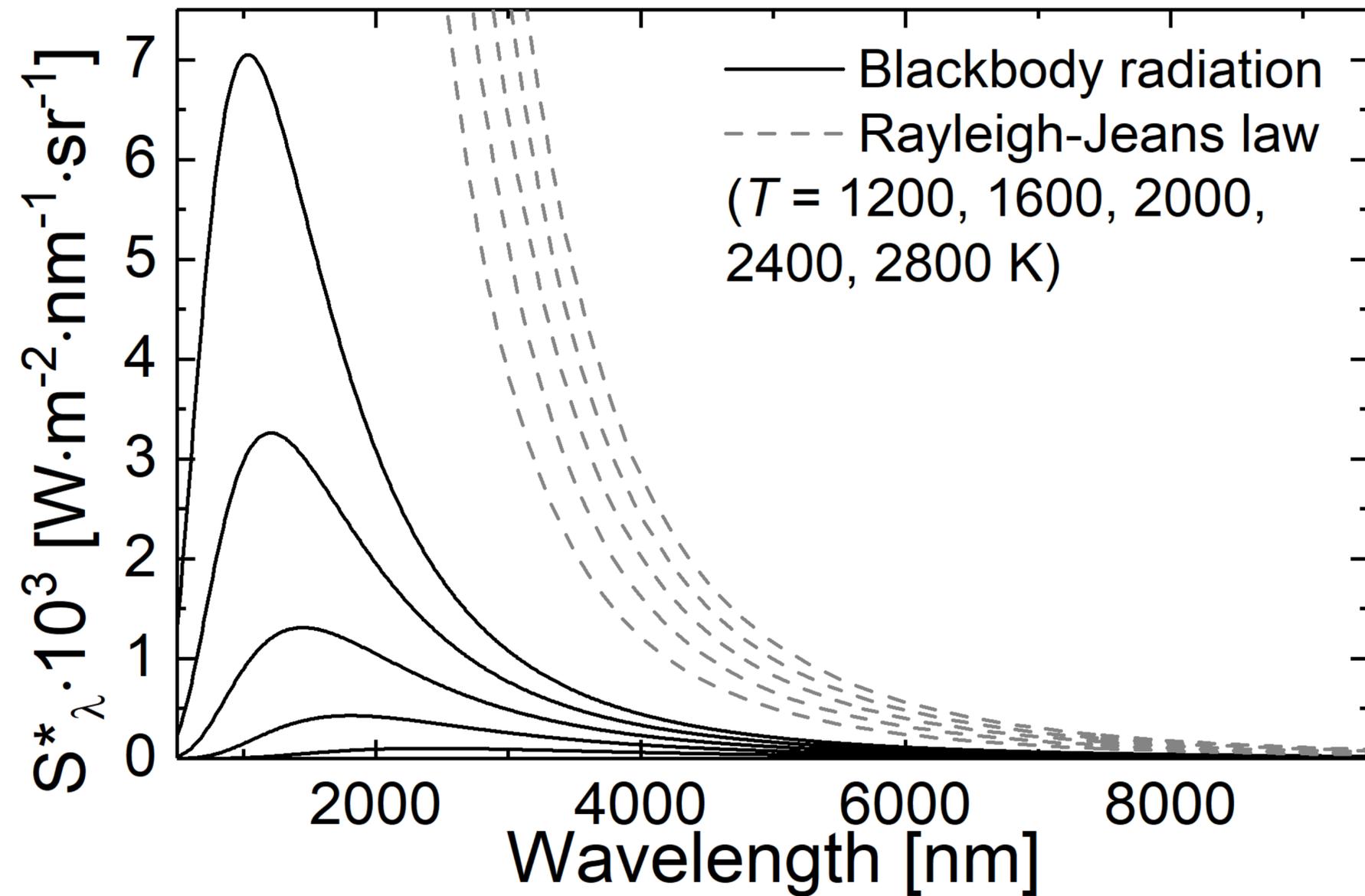
Cavity radiator and integrating or Ulbricht sphere



Wien's displacement law or Wien approximation



Rayleigh-Jeans law

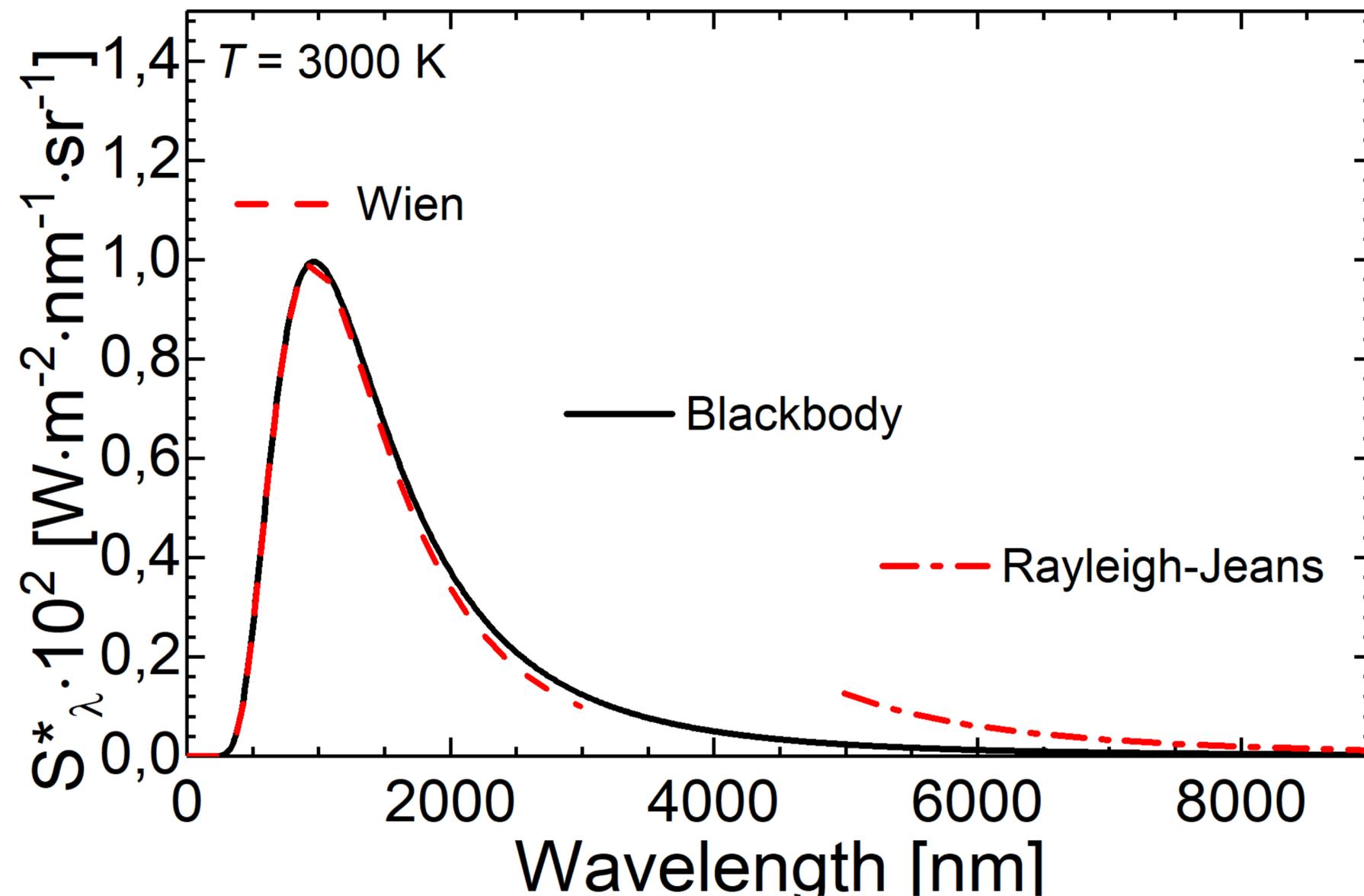


Correct representation of
- Long wavelength flank

Failed to describe
- Peak wavelength

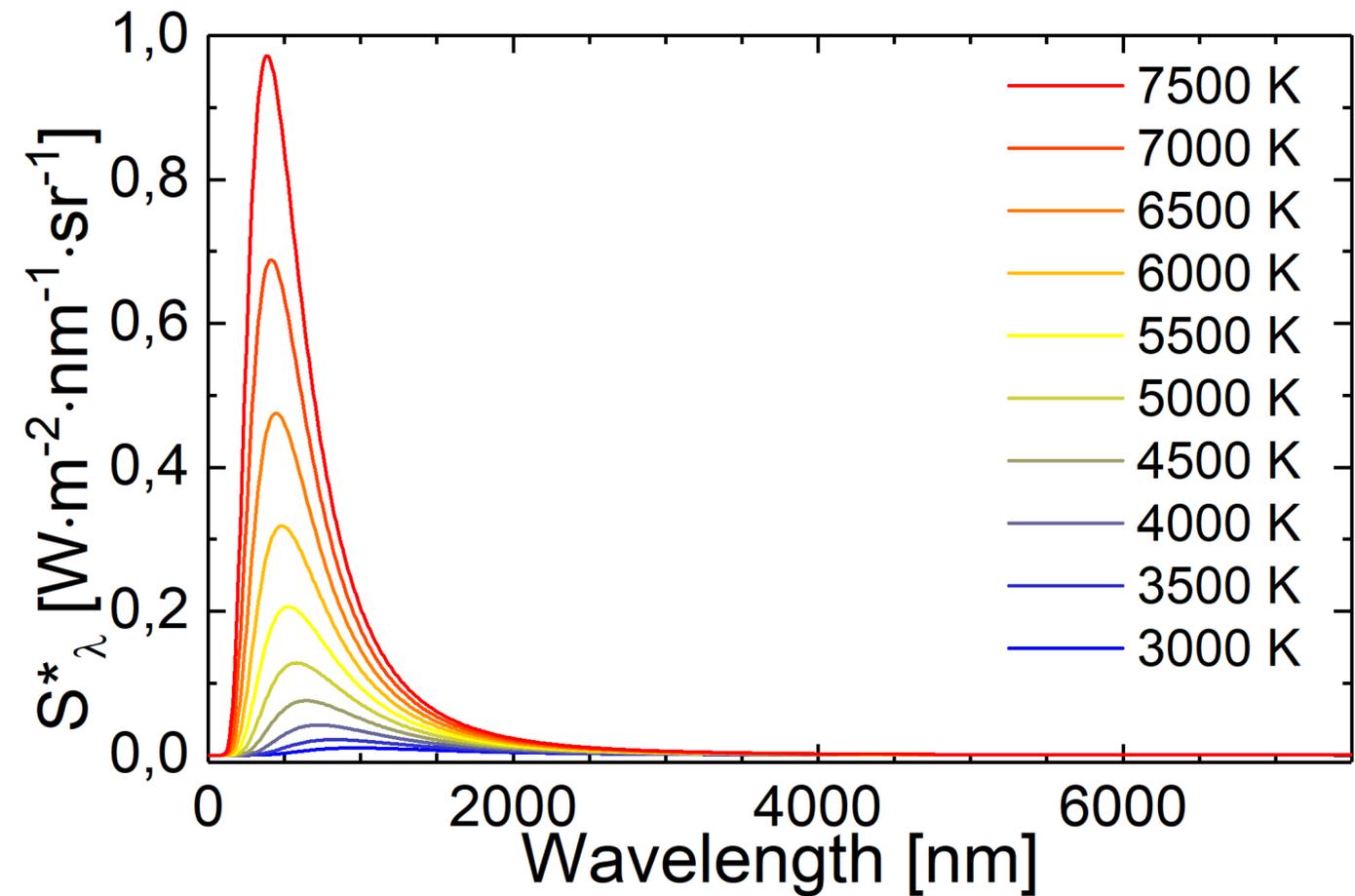
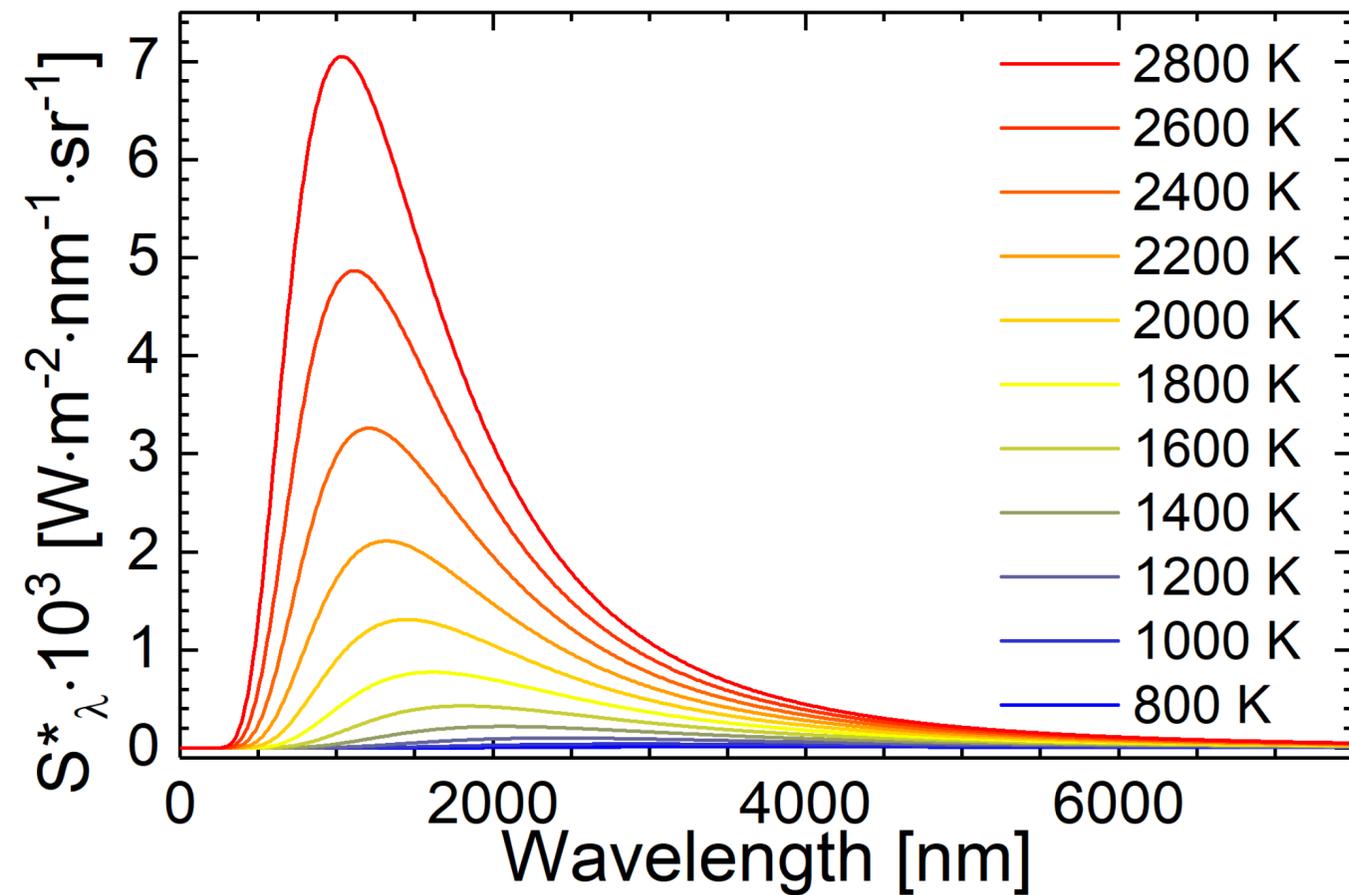
Leads to ultraviolet catastrophe

The gap between Wien approximation and Rayleigh-Jeans law

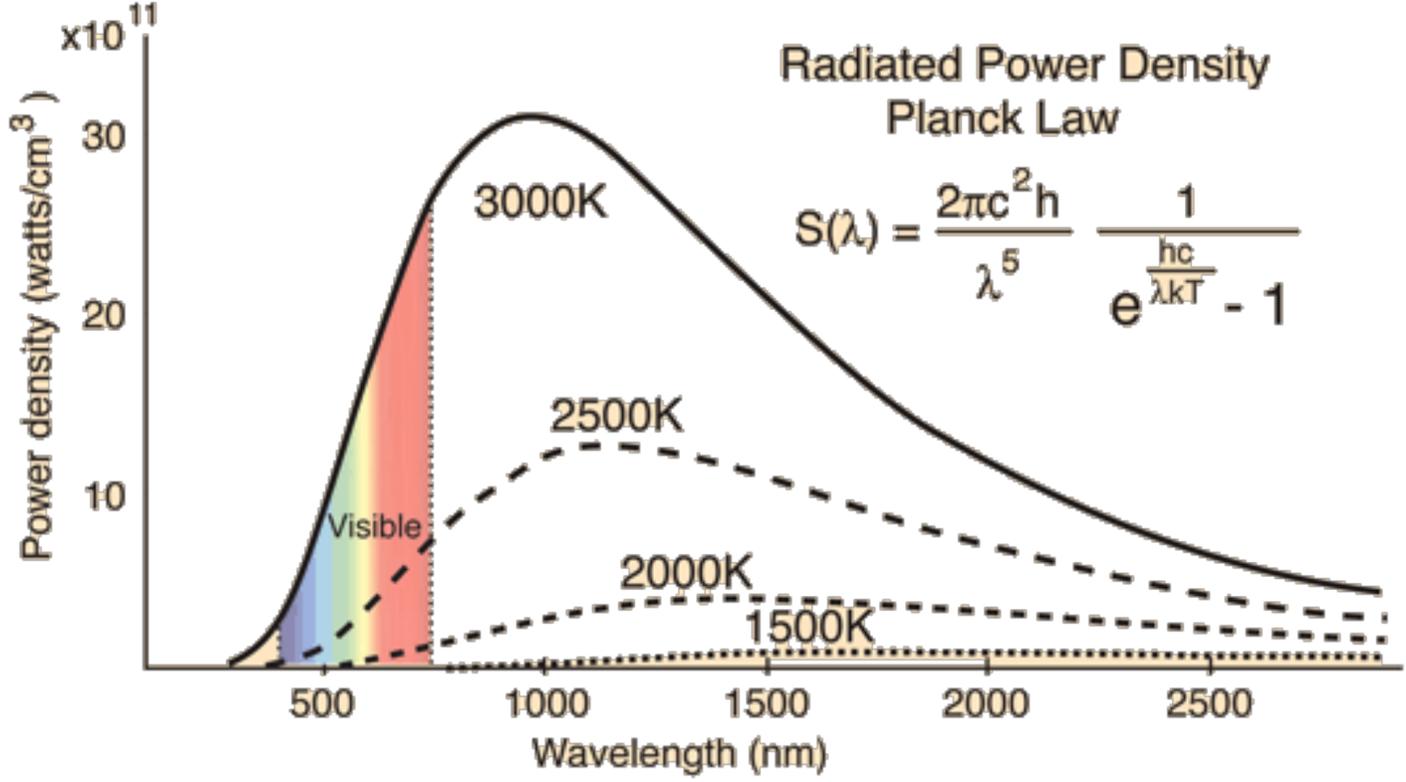
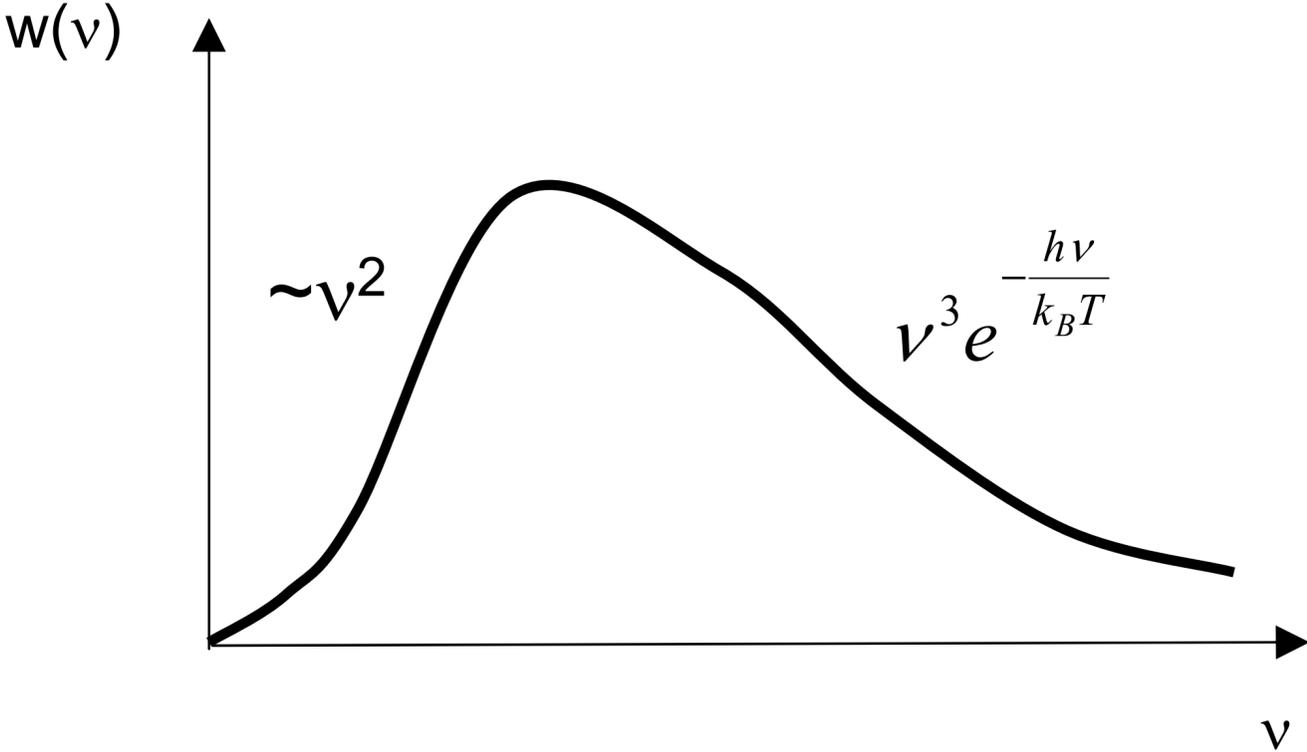


How to close the gap between *Wien approximation* that correctly describes the short wavelength flank and the *Rayleigh-Jeans law* that sufficiently describes the long wavelength flank?

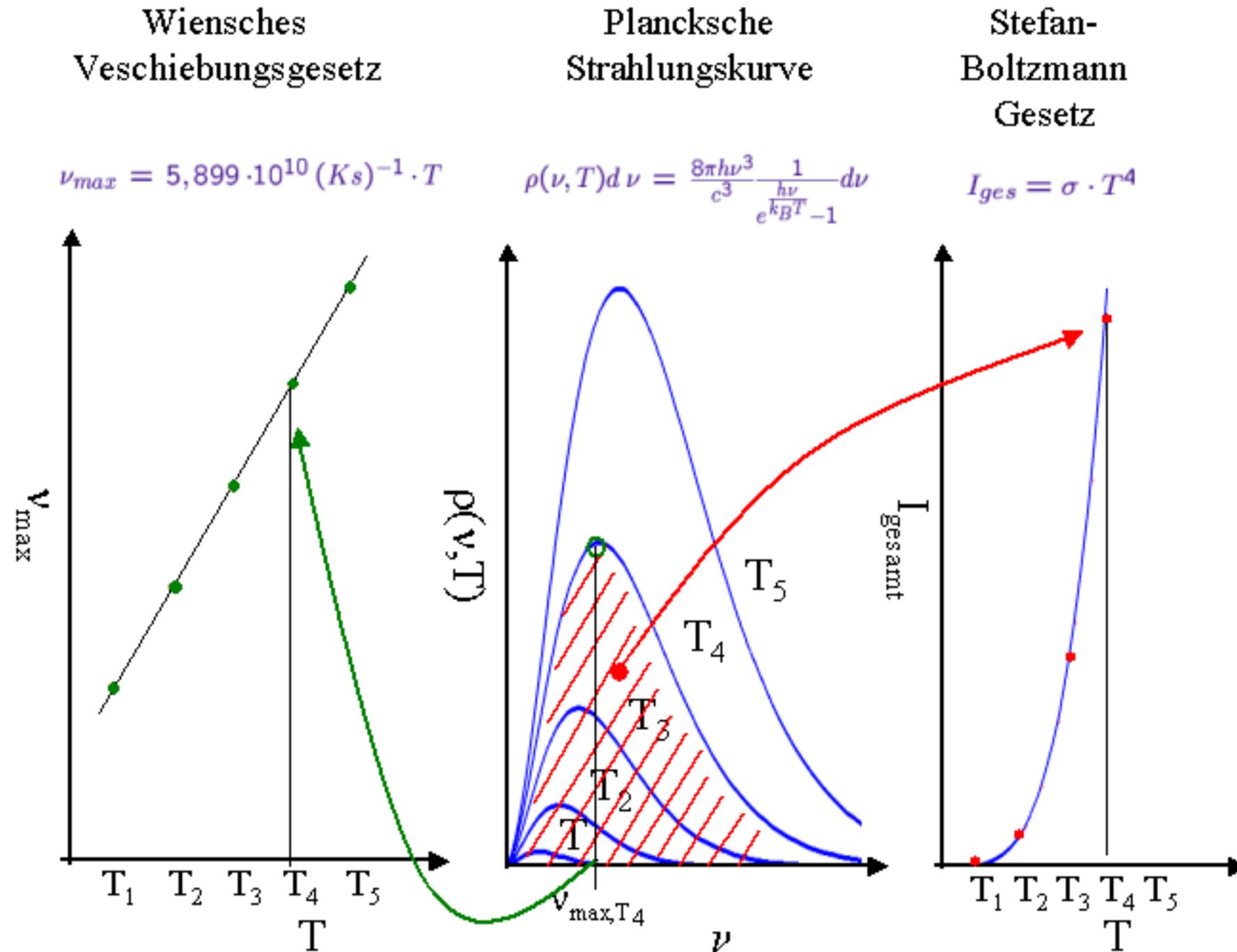
Planck's law of radiation



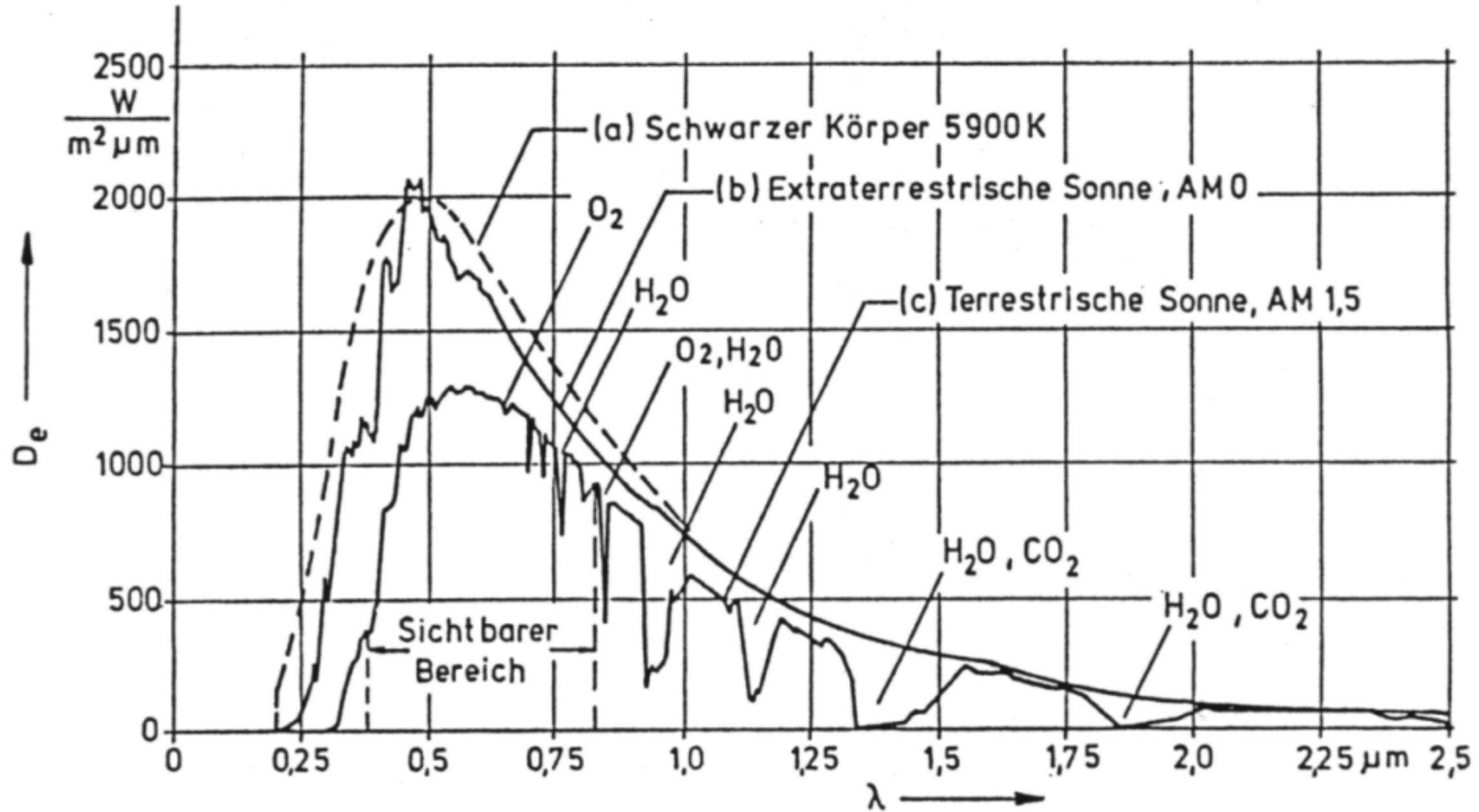
Planck's law of radiation



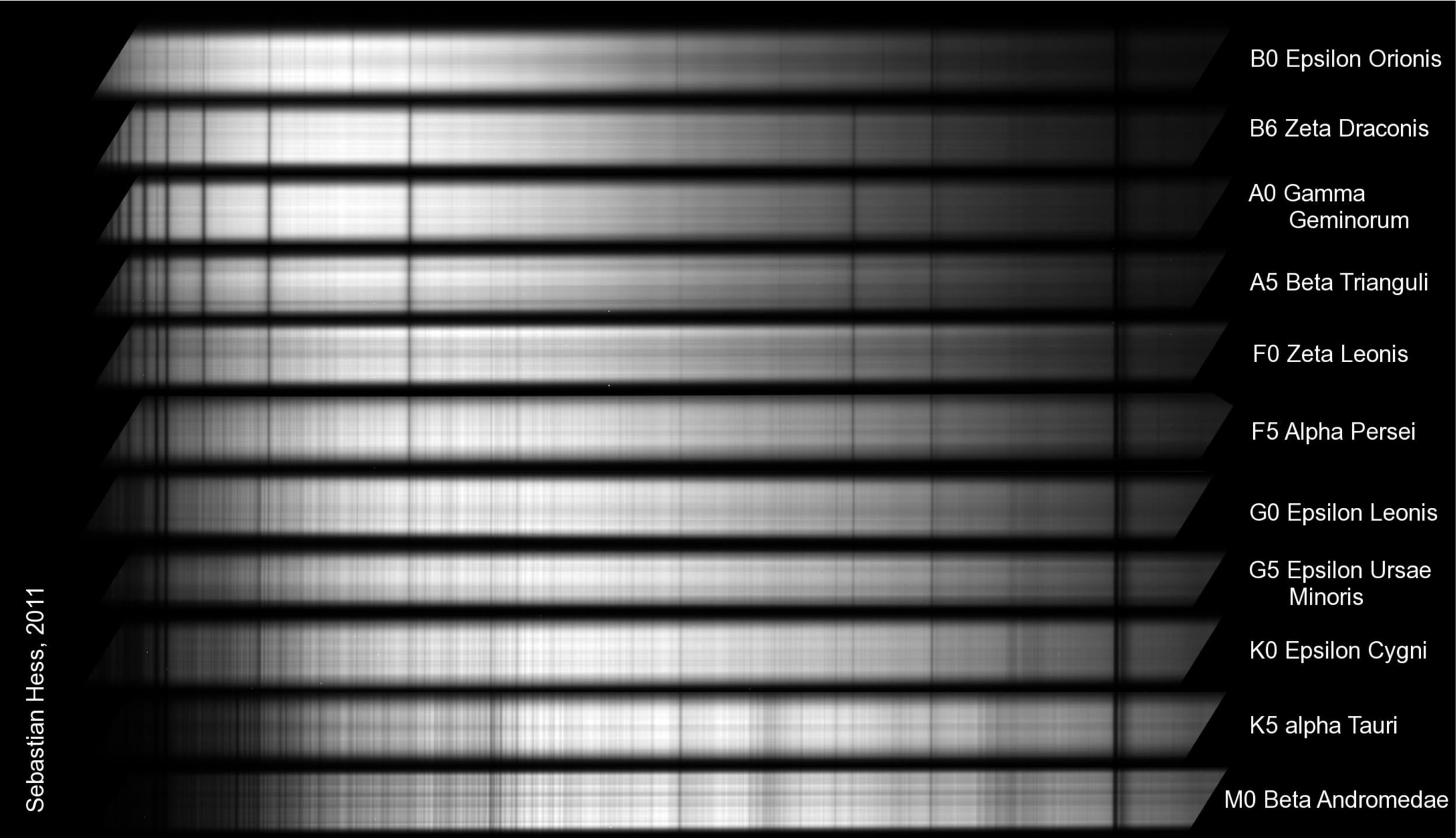
Planck's law of radiation



Electromagnetic spectrum of the sun



Electromagnetic spectrum of stars

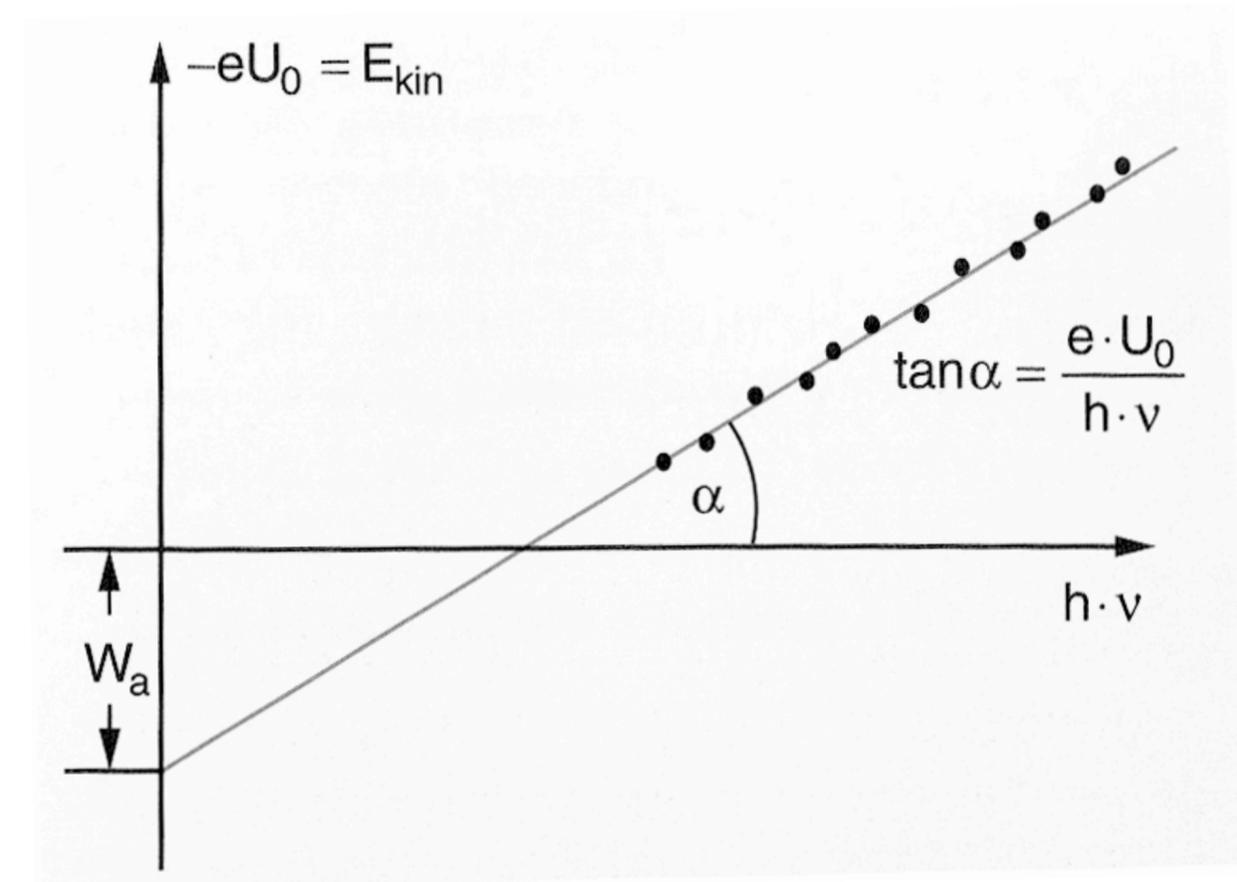
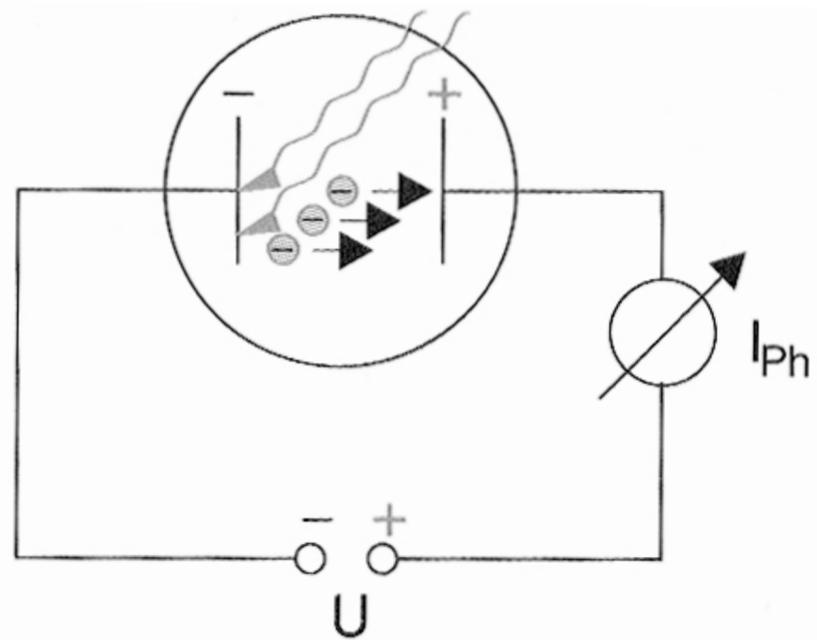


The Particle Nature of light

The photo law of radiation

Lennard 1902

- Kinetic energy of the photo electrons only depends on the wavelength of light
- Number of photo electrons only depends on the intensity
- Electrons are emitted without retardation



The inner photo law of radiation

